Chapter Ten

MECHANICAL PROPERTIES OF FLUIDS



10.1 A tall cylinder is filled with viscous oil. A round pebble is dropped from the top with zero initial velocity. From the plot shown in Fig. 10.1, indicate the one that represents the velocity (v) of the pebble as a function of time (t).

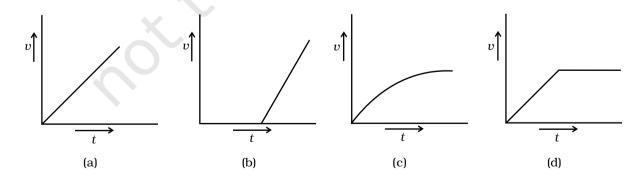
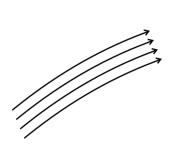
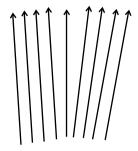
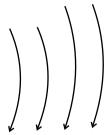


Fig. 10.1

10.2 Which of the following diagrams (Fig. 10.2) does not represent a streamline flow?







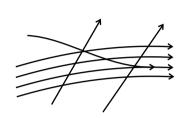


Fig. 10.2

- **10.3** Along a streamline
 - (a) the velocity of a fluid particle remains constant.
 - (b) the velocity of all fluid particles crossing a given position is constant.
 - (c) the velocity of all fluid particles at a given instant is constant.
 - (d) the speed of a fluid particle remains constant.
- 10.4 An ideal fluid flows through a pipe of circular cross-section made of two sections with diameters 2.5 cm and 3.75 cm. The ratio of the velocities in the two pipes is
 - (a) 9:4
 - (b) 3:2
 - (c) $\sqrt{3}:\sqrt{2}$
 - (d) $\sqrt{2}:\sqrt{3}$
- 10.5 The angle of contact at the interface of water-glass is 0° , Ethylalcohol-glass is 0° , Mercury-glass is 140° and Methyliodide-glass is 30° . A glass capillary is put in a trough containing one of these four liquids. It is observed that the meniscus is convex. The liquid in the trough is
 - (a) water
 - (b) ethylalcohol
 - (c) mercury
 - (d) methyliodide.

MCQ II

- **10.6** For a surface molecule
 - (a) the net force on it is zero.
 - (b) there is a net downward force.

73

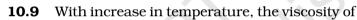


Exemplar Problems-Physics

- (c) the potential energy is less than that of a molecule inside.
- (d) the potential energy is more than that of a molecule inside.
- 10.7 Pressure is a scalar quantity because
 - (a) it is the ratio of force to area and both force and area are vectors.
 - (b) it is the ratio of the magnitude of the force to area.
 - (c) it is the ratio of the component of the force normal to the area.
 - (d) it does not depend on the size of the area chosen.
- 10.8 A wooden block with a coin placed on its top, floats in water as shown in Fig. 10.3.

The distance l and h are shown in the figure. After some time the coin falls into the water. Then

- (a) 1 decreases.
- (b) h decreases.
- (c) lincreases.
- (d) h increase.



- (a) gases decreases.
- (b) liquids increases.
- (c) gases increases.
- (d) liquids decreases.
- 10.10 Streamline flow is more likely for liquids with
 - (a) high density.
 - (b) high viscosity.
 - (c) low density.
 - (d) low viscosity.

VSA

- **10.11** Is viscosity a vector?
- **10.12** Is surface tension a vector?
- 10.13 Iceberg floats in water with part of it submerged. What is the fraction of the volume of iceberg submerged if the density of ice is ρ_i = 0.917 g cm⁻³?
- **10.14** A vessel filled with water is kept on a weighing pan and the scale adjusted to zero. A block of mass M and density ρ is suspended by a massless spring of spring constant k. This block is submerged inside into the water in the vessel. What is the reading of the scale?

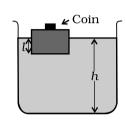


Fig. 10.3

74



10.15 A cubical block of density ρ is floating on the surface of water. Out of its height L, fraction x is submerged in water. The vessel is in an elevator accelerating upward with acceleration a. What is the fraction immersed?

SA

- **10.16** The sap in trees, which consists mainly of water in summer, rises in a system of capillaries of radius $r = 2.5 \times 10^{-5}$ m. The surface tension of sap is $T = 7.28 \times 10^{-2}$ Nm⁻¹ and the angle of contact is 0°. Does surface tension alone account for the supply of water to the top of all trees?
- **10.17** The free surface of oil in a tanker, at rest, is horizontal. If the tanker starts accelerating the free surface will be titled by an angle θ . If the acceleration is a m s⁻², what will be the slope of the free surface?.
- **10.18** Two mercury droplets of radii 0.1 cm. and 0.2 cm. collapse into one single drop. What amount of energy is released? The surface tension of mercury $T=435.5\times10^{-3}\,\mathrm{N}\;\mathrm{m}^{-1}$.
- **10.19** If a drop of liquid breaks into smaller droplets, it results in lowering of temperature of the droplets. Let a drop of radius R, break into N small droplets each of radius r. Estimate the drop in temperature.
- **10.20** The sufrace tension and vapour pressure of water at 20° C is $7.28 \times 10^{-2} \, \text{Nm}^{-1}$ and $2.33 \times 10^{3} \, \text{Pa}$, respectively. What is the radius of the smallest spherical water droplet which can form without evaporating at 20° C?

LA

- **10.21** (a) Pressure decreases as one ascends the atmosphere. If the density of air is ρ , what is the change in pressure d*p* over a differential height d*h*?
 - (b) Considering the pressure p to be proportional to the density, find the pressure p at a height h if the pressure on the surface of the earth is p_0 .
 - (c) If $p_0 = 1.03 \times 10^5$ N m⁻², $\rho_0 = 1.29$ kg m⁻³ and g = 9.8 m s⁻², at what height will the pressure drop to (1/10) the value at the surface of the earth?
 - (d) This model of the atmosphere works for relatively small distances. Identify the underlying assumption that limits the model.
- **10.22** Surface tension is exhibited by liquids due to force of attraction between molecules of the liquid. The surface tension decreases

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with increase in temperature and vanishes at boiling point. Given that the latent heat of vaporisation for water $L_{_{D}}=540~\rm k$ cal kg $^{-1}$, the mechanical equivalent of heat J = 4.2 J cal $^{-1}$, density of water $\rho_{_{\rm w}}=10^3~\rm kg$ t^{-1} , Avagadro's No $N_{_{\rm A}}=6.0\times10^{26}~\rm k$ mole $^{-1}$ and the molecular weight of water $M_{_{\rm A}}=18~\rm kg$ for 1 k mole.

- (a) estimate the energy required for one molecule of water to evaporate.
- (b) show that the inter-molecular distance for water is

$$d = \left[\frac{M_{\rm A}}{N_{\rm A}} \times \frac{1}{\rho_w}\right]^{1/3}$$
 and find its value.

- (c) 1 g of water in the vapor state at 1 atm occupies 1601cm³. Estimate the intermolecular distance at boiling point, in the vapour state.
- (d) During vaporisation a molecule overcomes a force F, assumed constant, to go from an inter-molecular distance d to d'. Estimate the value of F.
- (e) Calculate F/d, which is a measure of the surface tension.
- **10.23** A hot air balloon is a sphere of radius 8 m. The air inside is at a temperature of 60°C. How large a mass can the balloon lift when the outside temperature is 20°C? (Assume air is an ideal gas, R = 8.314 J mole⁻¹K⁻¹, 1 atm. = 1.013×10^5 Pa; the membrane tension is 5 N m⁻¹.)